Surname	Centre Number	Candidate Number
First name(s)		0



GCSE

3440U20-1



TUESDAY, 17 MAY 2022 - MORNING

APPLIED SCIENCE (Single Award) UNIT 2: Science to Support our Lifestyles

FOUNDATION TIER

1 hour 30 minutes

For Exa	For Examiner's use only						
Question	Maximum Mark	Mark Awarded					
1.	5						
2.	8						
3.	8						
4.	8						
5.	11						
6.	7						
7.	9						
8.	11						
9.	8						
Total	75						

ADDITIONAL MATERIALS

In addition to this paper you will require a calculator.

INSTRUCTIONS TO CANDIDATES

Use black ink or black ball-point pen. Do not use gel pen or correction fluid.

You may use a pencil for graphs and diagrams only.

Write your name, centre number and candidate number in the spaces at the top of this page.

Answer **all** questions.

Write your answers in the spaces provided in this booklet.

INFORMATION FOR CANDIDATES

The number of marks is given in brackets at the end of each question or part-question.

Question **7**(b) is a quality of extended response (QER) question where your writing skills will be assessed.

The Periodic Table is printed on page 20 of this examination paper.



Answer all questions.

- 1. There have been many advances in medical science since the discovery of the structure of DNA in the 1950s.
 - (a) Select the correct word or words from the list below to complete each of the following sentences.

genes	infra-red	DNA	ionising	Down's syndrome	cystic fibrosis

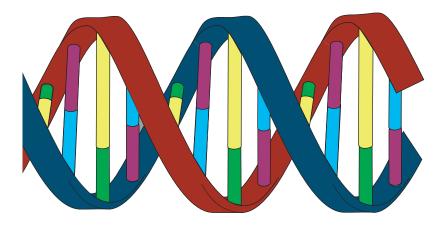
	(i)	Sections of DNA	that determine	inherited	characteristics	are	called
--	-----	-----------------	----------------	-----------	-----------------	-----	--------

(ii) Chromosomes contain [1]

(iii) Exposure toradiation can increase the chance of mutation in our DNA.

(iv) Too many copies of one particular chromosome causes

(b) DNA has a ladder-like, double helix structure with bases (A, C, G and T) forming the rungs.



Complete the table below to show how the bases are paired.

[1]

Base	Pair
А	
С	



Jennie has type 1 diabetes. She monitors her blood glucose levels using an electronic meter.



Jennie's results for one day are given in the table below.

Meal	Glucose level before eating (units)	Glucose level two hours after eating (units)	Glucose levels within range (\(\setminus / \times \)
breakfast	90	134	
lunch	95	152	
dinner	66	58	
supper	92	126	

- Complete the data in the table above to show whether the blood glucose levels are within (a) the target range of 70–140 two hours after each meal.
- (b) After eating dinner, Jennie's glucose level is below the target range. State what Jennie will need to do to bring her level within range.

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	ii) Suggest how .	Jennie's uncle can control his	diabetes.	[1
•···				
	heir results are sho	tested some artificial urine sale with the table below. Colour of reagent before		glucose.
			Colour of reagent after testing	glucose.
	heir results are sho	own in the table below. Colour of reagent before	Colour of reagent after	glucose.
	heir results are sho	Colour of reagent before testing	Colour of reagent after testing	glucose.
	heir results are sho	Colour of reagent before testing	Colour of reagent after testing amber	glucose.
	heir results are sho	Colour of reagent before testing blue blue	Colour of reagent after testing amber green	glucose.
Ti	Sample A B C D	Colour of reagent before testing blue blue blue blue blue blue blue	Colour of reagent after testing amber green blue	
Ti	Sample A B C	Colour of reagent before testing blue blue blue blue blue blue blue	Colour of reagent after testing amber green blue	glucose. [1

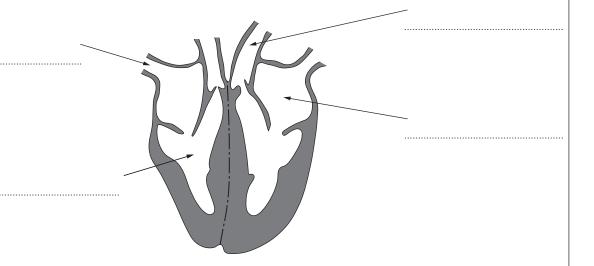
Examiner only

The diagram below shows a cross-section through a hea	eart.	a h	through	cross-section	shows a	below	The diagram	3.
---	-------	-----	---------	---------------	---------	-------	-------------	----

(a) Label the diagram using the words in the box below.

[4]

vein atrium ventricle artery valve



 (b) The diagrams in the table below show the relative thickness of the walls of three different blood vessels.
 Complete the table.

[4]

Diagram	Name	Description	Purpose
	Capillary	walls are only one cell thick	
			carries blood at a high pressure
		thin walls with valves	



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3440U201

4. Public Health Wales is encouraging people over the age of 50 to 'Steady on ... Stay SAFE' to reduce their risk of falling.

Falls are common, and a serious health issue for older people. Around a third of all people aged 65 fall at least once each year. This increases to 50% for those aged 80 and over.

Around 5% of falls lead to fractures of joints. A common fracture in people over 50 is a broken hip.

The table below shows the results of a study into the number of hip fractures at different ages.

Ago	Number of hip fractures per year				
Age	per 10 000 men	per 10 000 women			
30	<1	<1			
40	<1	<1			
50	3	3			
60	5	10			
70	30	50			
80	80	130			

(2)	Ancwortho	following	ucina	tho	inform	ation	in the	table
(a)	Answer the	tollowing	usina	tne	Intorm	ation	in the	table.

(i)	State the number	of cases of hi	n fracture nei	vear in women	at 70 v	ears of age	[1]
(1)	Otate the number	or cases or m	p macture per	year in wonten	atroy	cars or ago	· [']

number of cases = per 10 000

(ii)	There are approximately 500000 women in the UK aged 70.	
	Estimate the total number of women aged 70 who are likely to break a hip during	
	a year.	2

number of women aged 70 that break a hip =



	(iii)	A health magazine states that a 67-year-old man is at less risk of a hip fracture than a 60-year-old woman. Explain whether you agree or disagree with the magazine. The state of t		Examı only
		II. State how this study could be improved to obtain a more definite conclusion for your answer above.	on [1]	
(b)		r people sometimes need a replacement hip joint. plete the following sentences by underlining the correct word(s) in the brackets.		
	(i)		[1]	
	(ii)	Older people sometimes need a hip replacement due to (osteoarthritis / diabetes / heart disease).	[1]	
				8

- 5. Technetium-99m is a common radioactive tracer used in gamma camera imaging. It cannot be stored for very long because it has a short half-life of 6 hours. It is extracted from molybdenum-99 when required. Molybdenum-99 is a beta (β) emitter with a much longer half-life that allows it to be stored for up to seven days.
 - (a) Complete the decay equation for molybdenum-99.

[2]

$$^{99}_{42}\text{Mo} \rightarrow ^{99}\text{Tc} + \dots$$

(b) A sample of technetium-99m injected into a patient consists of 10 000 nuclei. The table below shows the number of nuclei remaining after each half-life.

Time (hours)	Number of technetium-99m nuclei
0	10 000
6	5000
12	2500
18	
24	625
30	313
36	157

(i) Complete the table.

[1]

Examiner

Plot a graph on the grid below to show how the number of technetium-99m nuclei (ii) changes over a 36 hour period. Draw a suitable line.

Number of technetium-99m nuclei

Molybdenum-99 has a much longer half-life than technetium-99m. (iii)

Sketch another line on the grid to show the likely decay of molybdenum-99 if there were 10 000 nuclei at 0 hours. [2]

State two precautions that should be taken when using technetium-99m in gamma (c) camera imaging.

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[2]

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[2]

6.	Wales has a long history of mining metal ores. Leakage from disused mines is a significant
	source of water pollution today.

State **two** symptoms caused by this exposure.

Frongoch mine in Ceredigion produced lead and zinc ore during the 18th and 19th centuries. The Frongoch stream flows through the mine and into the River Ystwyth.

(a)	Long-term	exposure to	heavy n	netals such	as zinc and	lead can	damage	human health.
-----	-----------	-------------	---------	-------------	-------------	----------	--------	---------------

1.

(b) In 2011, Natural Resources Wales diverted Frongoch stream so it no longer flowed through the mine.

They collected the following data on the time taken for water to flow to the River Ystwyth.

	Distance stream flows (m)	Time taken for water to reach the River Ystwyth (s)	Volume of water entering the River Ystwyth per second (dm³/s)
Before Diversion	500	800	17
After Diversion	2000	4000	87

A local newspaper article stated that, following the diversion, the speed of the water flowing had increased.

(i) Use the equation

$$\frac{\text{speed} = \frac{\text{distance}}{\text{time}}}$$

to explain whether the newspaper was correct.

[4]

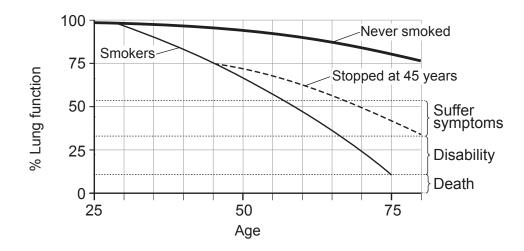
(ii)	The newspaper also claimed that as the speed of the water increased the volume of water per second entering the River Ystwyth also increased. Explain whether you agree.	Examine only
		7



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7.	The	symptoms of lung disease are shortness of breath, wheezing and constant coughir	ıg.
		ldy has been conducted into the effect of smoking on lung function. The lung function different groups of people was measured:	on of
	•	people who smoke people who did smoke but gave up smoking at the age of 45 people who have never smoked.	
	(a)	A large sample of people is used in this type of study.	
		Tick (✓) the three correct statements below.	[3]
		A large number of people are studied to decrease the correlation of the results.	
		A large number of people are studied to improve the confidence in the results.	
		A result that does not fit the pattern is called an outlier.	
		The conclusion must be incorrect if there is one outlier.	
		Correlation is a measure of how one variable relates to another.	
		A smaller number of people studied provide better evidence.	

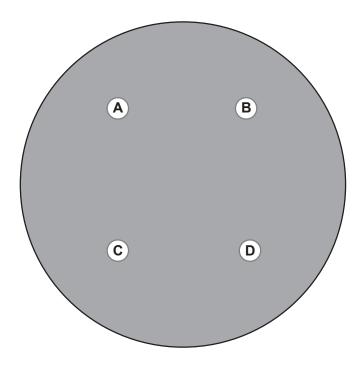
(b) The results of the study are shown in the graph below.



Use the graph to explain how lung function changes with age and how this affects each of the three groups of people studied. [6 QER]

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8. A group of students investigated how different antibiotics affected the growth of bacteria. An agar plate was inoculated with bacteria and small discs containing different antibiotics (**A**, **B**, **C** and **D**) were placed onto it.



The following procedure was followed to add the antibiotic discs to an agar plate inoculated with bacteria.

- 1. Pick up the filter paper disc containing antibiotic **A** using forceps.
- 2. Place the antibiotic disc on to the plate.
- 3. Repeat steps 1 and 2 for antibiotic discs B, C and D.
- 4. Place a lid on the agar plate.
- 5. Turn the plate upside down and incubate for 3 days at 20°C.
- 6. Record the radius of the clear zone around each disc.

(a)	State two wa	ays the students	could reduce t	the contamination	on by unwanted	l bacteria.	[2]
• • • • • • • • • • • • • • • • • • • •							· · · · · ·



Examiner only

(b) The students obtained the following results.

Antibiotic disc	Radius of clear zone (cm)
Α	1.2
В	0.7
С	1.4
D	0.2

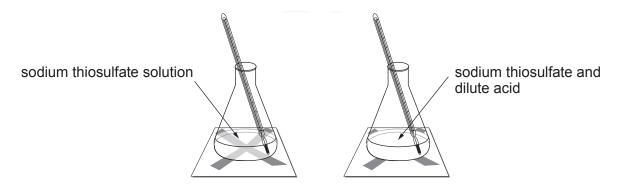
(i)	Explain which antibiotic is least effective.	[2]
(ii)	Use the equation	
	area = $3.14 \times \text{radius}^2$	
	to calculate the area of the clear zone for antibiotic C to three significant figures	s. [3]
	area = c	cm²
(iii)	It has been estimated that each 1 cm ² on this plate contains 50 000 bacteria.	
	Calculate how many bacteria were present in the area cleared by antibiotic C at the start of the experiment.	t [2]
	number of bacteria =	
(iv)	Robert says that antibiotic C is twice as effective as antibiotic B . Jeremy says that antibiotic C is four times as effective as antibiotic B . Explain whether you agree with Jeremy or Robert.	[2]
		······•

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9. A student carried out an experiment to investigate how concentration affects the rate of a chemical reaction.

When hydrochloric acid (HCI) reacts with sodium thiosulfate solution a precipitate of sulfur is formed that makes the solution go cloudy.

The student made different concentrations of hydrochloric acid by adding water to hydrochloric acid. These were then added to a sodium thiosulfate solution at 20 °C. The time for the cross to disappear was then measured.



The following results were obtained.

Volume of HCI (cm ³)	Volume of water (cm ³)	Time for cross to disappear (s)
25	0	2
20	5	9
15	10	37
10	15	75
5	20	153

(a)	State two variables that need to be controlled in this experiment.	
	1	
	2	
(b)	Explain the results in terms of particles.	[3]
•••••		
•••••		



	17	— Eva-
(C)	Explain, in terms of particles, what would happen to the results if the experiment was carried out at $5^{\circ}\text{C}.$	Exam only
•••••		
	END OF PAPER	8



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Heilum Heilum 2 20 Neon 10 Neon 10 10 Neon 10

Nitrogen
7
31
Phosphorus
15
As
Arsenic
33
122
Sb
Antimony
51
209
Bi
Bismuth
83

Zinc Zinc 30 30 112 Cd 348 48 Hg Mercury 80

63.5 Cu Copper 29 29 108 Ag Silver 47 197 Au Gold 79

Nickel 28 28 106 Pd 46 195 Pt 195 Pt 195 Pt 78 78

Co Cobalt 27 103 Rhodium 45 192 Ir

Fe Fe Iron 26 101 Ruthenium 44 4 190 Os Osmium 76

Mn Manganese 25 99 Tc Tchretium 43 186 Re

Ti Ti Ti 22 22 91 Zirconium 40 179 Hf Hafnium

(3440U20-1)

Scandium
Scandium
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21
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Y
Yttrium
39
139
La
Lanthanum
57
227
Ac
Actinium
89

Chromium 24 96 Mo Molybdenum 42 184 W Tungsten 74

Boron 5 Aluminium Aluminium 31 70 Ga Gallium 31 115 In Indium 49 204 TI

· relative atomic mass

Key

. atomic number